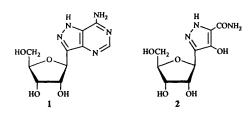
Stereoselective synthesis of 3- β -D-ribofuranosylpyrazole from 2,3-O-Isopropylidene-D-ribose; a new route to pyrazole C-nucleosides

Anthony D. Rycroft," Gurdial Singh *." and Richard H. Wightman *.b

^a School of Science and Technology, University of Teesside, Middlesbrough, Cleveland TS1 3BA, UK ^b Department of Chemistry, Heriot-Watt University, Riccarton, Edinburgh EH14 4AS, UK

2,3-O-Isopropylidene-D-ribose has been converted stereoselectively into $3(5)-(\beta$ -D-ribofuranosyl)pyrazole 11, a precursor for the C-nucleoside antibiotics pyrazofurin and formycin.

Since the isolation of the C-nucleoside antibiotics, there has been considerable interest in synthetic approaches to these compounds, including in particular the pyrazole C-nucleosides formycin 1 and pyrazofurin 2.¹ Previous work in one of our



laboratories has led to the development of routes to pentofuranosylpyrazoles using acetylenic intermediates, and we have shown that the ribofuranosylpyrazole 3 can be efficiently converted in one step into the 1,4-dinitropyrazole 4 (Scheme 1) by treatment with NH4NO3, trifluoroacetic acid and trifluoroacetic anhydride,² and that 4 can be transformed, also in one high-yielding step, to the doubly functionalized pyrazole 5 using NaCN in aqueous ethanol.³ Formycin 1³ and pyrazofurin 2⁴ can then be obtained from 4. The synthesis of 3, however, was not totally satisfactory, depending on the addition of an acetylenic Grignard reagent to 2,3,5-tri-O-benzyl-D-ribofuranose which proceeds with somewhat low stereoselectivity.⁵ We here describe a new and more direct synthesis of intermediate 3 from 2,3-O-isopropylidene-D-ribofuranose, a derivative of ribose which is much more easily prepared than is the tribenzyl ether, and which relies upon the stereoselective reduction of an acetylenic hemiketal.

The lactone 6 (Scheme 2) is easily prepared from 2,3-Oisopropylidene-D-ribofuranose⁶ by silylation followed by oxidation using KMnO₄. Treatment of 6 with the lithium derivative of 1,1-diethoxyprop-2-yne gave the hemiketal 7 (85%), which on reduction with NaBH₄ gave the *syn-* (*threo-*) diol 8, as an inseparable 6:1 mixture with its diastereoisomer (90% combined yield). The diastereoselectivity in this reduction is in the sense predicted from a Felkin–Anh model⁷ for the transition state; we have recently described a similar stereoselectivity in the reduction of the adduct from 6 and allylmagnesium chloride,⁸ and other related hemiacetals behave similarly.⁹ Treatment of this mixture of diols with TsCl in pyridine led, by selective sulfonation at the prop-2-ynlic alcohol,⁵ to the formation of the β -alkyne 9 (52%), together with the α -anomer (10%); a third product of β -L-lyxo-configuration was also formed (7%) as a result of sulfonation at the alternative alcohol.[†] The structure of the major product 9 was supported by the observation of small values (2.85 and 1.85 Hz respectively) for the coupling constants $J_{4.5}$ and $J_{6.7}$, and the observation of strong NOE effects between 4-H and 7-H. When 9 was treated with acetic acid and aqueous HCl, followed by hydrazine hydrate, the pyrazolyl diol 10 was obtained. This on desilylation gave 3(5)-(β -D-ribosuranosyl)pyrazole 11, which could be acetylated to produce the tetra-acetyl compound 3, with spectroscopic and optical properties {[α]_D - 10.9 (c 1.28 CHCl₃)},[‡] in excellent agreement with those reported {[α]_D - 9.9 (c 3.4, CHCl₃)}.²

Our findings described here, together with earlier reports of the efficient conversion of **3** into formycin and pyrazofurin,²⁻⁴ offer an effective route to these antibiotics. The strategy used is likely to be of value in gaining stereocontrolled access to other *C*-glycosides and *C*-nucleosides.

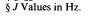
Experimental

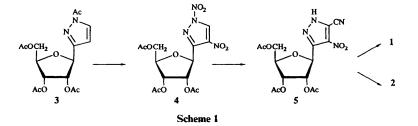
8-*O-tert*-Butyldiphenylsilyl-2,2,3,3-tetradehydro-2,3-dideoxy-5,6-*O*-isopropylidene-D-*altro*-octose diethyl acetal 8

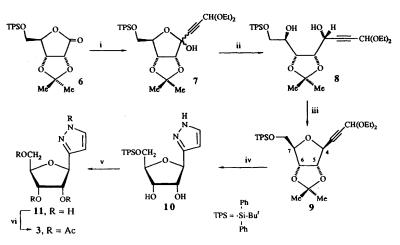
Sodium boranuide (1.07 g) was added in portions over 10 min to a stirred solution of the hemiacetal 7 (1.57 g) in methanol (75 cm³) at 0 °C. After 1 h, the mixture was evaporated and the residue was partitioned between water (120 cm³) and ethyl acetate (4 × 100 cm³). The organic layers were dried (Na₂SO₄) and evaporated to give an oil which was chromatographed on silica, with light petroleum-diethyl ether (7:3) as eluent to give the *diol* 8 (1.42 g, 90%) as a 6:1 mixture with the D-*allo*-epimer, $[\alpha]_D - 10.9$ (c 2.94, CHCl₃); $\delta_H(200 \text{ MHz}, \text{CDCl}_3)$ 1.07 (9 H, s), 1.20 (6 H, t), 1.38 (6 H, s), 3.05 (1 H, d, J 5.3),§ 3.4–3.95 (7 H,

† The structures of the minor products were supported by NMR data, in particular the progressive upfield shifts of C-4, -5, -6 and -7 in the series 9, α -anomer, β -L-*lyxo*-isomer, as would be expected on the basis of steric factors.

 $\ddagger [\alpha]_D$ values in 10^{-1} deg cm² g⁻¹.







Scheme 2 i, $LiC \equiv C-CH(OEt)_2$, $-78 \circ C$, 3 h; ii, NaBH₄, MeOH, 0 $\circ C$, 1 h; iii, TsCl, C_5H_5N , 90 $\circ C$, 25 h; iv, HOAc, HCl aq., room temp., then N₂H₄·H₂O, heat (55%); v, Bu₄NF, THF, room temp. (95%); vi, Ac₂O, C_5H_5N , DMAP, room temp., 5 h (95%)

m), 4.15–4.4 (3 H, m), 4.82 (1 H, ddd, J 1.0, 1.9, 3.8), 5.30 (1 H, br s), 7.4 (6 H, m) and 7.7 (4 H, m) [Found: MNH₄⁺ (CI, NH₃) 574.3200. Calc. for C₃₁H₄₈NO₇Si, 574.3200].

4,7-Anhydro-8-*O-tert*-butyldiphenylsilyl-2,2,3,3-tetradehydro-2,3-dideoxy-5,6-*O*-isopropylidene-D-*allo*-octose diethyl acetal 9

A solution of the diol 8(1.03 g) and toluene-*p*-sulforyl chloride (1.27 g) in dry pyridine (40 cm^3) was heated at 85–90 °C for 25 h. After cooling of the reaction mixture it was diluted with water (0.5 cm^3) and evaporated. Trituration of the residue with hot ether $(3 \times 30 \text{ cm}^3)$ and evaporation gave an oil which was chromatographed on silica, with light petroleum-diethyl ether (1:1) as eluent to give firstly the β -D-ribofuranosylalkyne 9 (0.52 g, 52%) as an oil, $[\alpha]_D$ – 5.5 (c 2.52, CHCl₃); δ_H (400 MHz, CDCl₃) 1.07 (9 H, s, CMe₃), 1.16 (6 H, dt, CH₂Me), 1.35 and 1.53 (each 3 H, s, CMe₂), 3.45–3.55 and 3.59–3.68 (each 2 H, m, CH₂Me), 3.78 (1 H, dd, J 10.8, 5.1, 8_a-H), 3.81 (1 H, dd, J 10.8, 6.6, 8_b-H), 4.19 (1 H, ddd, J6.7, 5.1, 1.85, 7-H), 4.68 (1 H, dd, J_{4.5} 2.9, ⁵*J*_{4,1} 1.4, 4-H), 4.75 (1 H, dd, *J*_{5.6} 6.2, *J*_{5.4} 2.85, 5-H), 4.78 (1 H, dd, J_{6,5} 6.2, J_{6,7} 1.85, 6-H), 5.15 (1 H, d, J_{1,4} 1.4, 1-H), 7.4 (6 H, m, Ph) and 7.7 (4 H, m, Ph); $\delta_{\rm C}(22.4 \text{ MHz}, \text{CDCl}_3)$ 19.2 (CMe₃), 14.9, 25.3 and 26.8 (each Me), 60.8 and 63.5 (CH₂), 74.5, 82.7, 85.8 and 86.1 (C-4-C-7), 82.0 and 82.9 (alkyne), 91.4 [CH(OEt)₂], 113.5 (CMe₂), 127.6, 129.7, 133.2 and 135.6 [Found: MNH₄⁺ (CI, NH₃) 556.3090. Calc. for C₃₁H₄₆NO₆Si, 556.30867.

Further elution of the column gave the α -D-ribofuranosylalkyne (0.106 g, 10%), as an oil, $[\alpha]_D - 35.0$ (*c* 0.8, CHCl₃); $\delta_H(200 \text{ MHz}, \text{CDCl}_3)$ 1.03 (9 H, s, CMe₃), 1.23 (6 H, dt, CH₂*Me*), 1.38 and 1.57 (each 3 H, s, CMe₂), 3.52–3.88 (6 H, m), 4.18 (1 H, t, *J* 2.8, 7-H), 4.85 (2 H, m), 5.05 (1 H, dd, *J* 3.5, 1.25, 4-H), 5.38 (1 H, d, *J* 1.25, 1-H), 7.4 (6 H, m, Ph) and 7.6 (4 H, m, Ph); $\delta_C(22.4 \text{ MHz}, \text{CDCl}_3)$ 73.8, 82.4, 83.1 and 84.3 (C-4–C-7).

Further elution gave the β -L-lyxo-isomer (0.074 g, 7%) as an oil, $[\alpha]_D - 34.5$ (c 0.52, CHCl₃); $\delta_H(200 \text{ MHz}, \text{ CDCl}_3)$ 1.06 (9 H, s, CMe₃), 1.22 (6 H, dt, CH₂Me), 1.33 and 1.40 (each 3 H,

s, CMe₂), 3.50–3.82 (5 H, m), 3.90 (1 H, dd, J 10.3, 5.8, 8_a -H), 3.99 (1 H, dd, J 10.3, 6.7, 8_b -H), 4.24 (1 H, dd, J 3.5, 1.2, 4-H), 4.72 (2 H, m, 2-H, 3-H), 5.33 (1 H, d, J 1.2, 1-H), 7.4 (6 H, m, Ph) and 7.7 (4 H, m, Ph); δ_c (22.4 MHz, CDCl₃) 72.5, 80.7, 81.7 and 82.1 (C-4–C-7).

Acknowledgements

We thank Professor J. G. Buchanan (University of Bath) for helpful discussions and encouragement, and EPSRC for a ROPA award (to G. S.) and for access to central facilities for mass spectrometry at the University of Wales, Swansea (Director, Dr J. A. Ballantine).

References

- 1 For a review, see J. G. Buchanan, Fortschr. Chem. Org. Naturst., 1983, 44, 243.
- 2 J. G. Buchanan, A. H. Jumaah, G. Kerr, R. R. Talekar and R. H. Wightman, J. Chem. Soc., Perkin Trans. 1, 1991, 1077.
- 3 J. G. Buchanan, A. Stobie and R. H. Wightman, *Can. J. Chem.*, 1980, **58**, 2624.
- 4 J. G. Buchanan, A. Stobie and R. H. Wightman, J. Chem. Soc., Perkin Trans. 1, 1981, 2374.
- 5 J. G. Buchanan, A. R. Edgar, R. J. Hutchison, A. Stobie and
- R. H. Wightman, J. Chem. Soc., Perkin Trans. 1, 1980, 2567.
- 6 N. A. Hughes and P. R. H. Speakman, Carbohydr. Res., 1965, 1, 171.
- 7 N. T. Anh, Top. Curr. Chem., 1980, 88, 145.
- 8 S. Jiang, B. Mekki, G. Singh and R. H. Wightman, *Tetrahedron Lett.*, 1994, **35**, 5505.
- 9 S. Jiang, G. Singh and R. H. Wightman, unpublished results.

Paper 5/05958E Received 8th September 1995 Accepted 14th September 1995